Q1.

Diagram 1 below shows the lungs and the trachea, the airway leading to the lungs. One of the lungs is drawn in section.

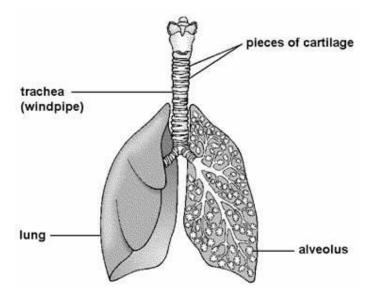


diagram 1

(a) In the wall of the trachea, there are pieces of a stiff material called car	tilage.
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why is this stiff material necessary in the wall of the trachea?	

1 mark

(b) Diagram 2 below shows one alveolus and its blood supply.

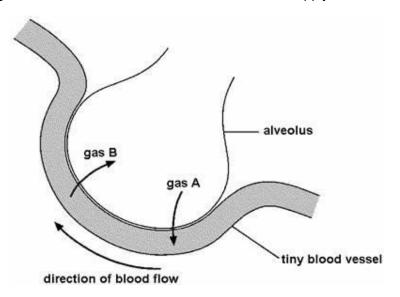


diagram 2

	(i)	Look at diagram 2, above. Gas A enters the blood from the alveolus. Gas B leaves the blood and enters the alveolus. What are the names of gases A and B?	
		gas A	
		gas B	1 mark
	(ii)	Give one reason why it is easy for gases to pass across the wall of an alveolus.	
			1 mark
(c)	The	diagram below shows a ciliated cell from the lining of the airway.	
	(i)	What is the function of this call in the circus?	
	(i)	What is the function of this cell in the airway?	
			1 mark
	(ii)	This cell is affected by substances in cigarette smoke. What effect does cigarette smoke have on the cilia?	
			4
	(iii)	Give the name of the substance, in cigarette smoke, which causes addiction to smoking.	1 mark
		Maximum 6	1 mark 3 marks

Q2.

The table shows the recommended daily intake of energy and some of the nutrients needed by different groups of people.

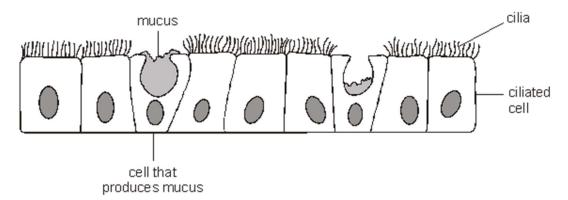
		nutrients				
group of	energy,	energy, protein, in ca	carbohydrate, in g	fat, in g	minerals, in g	
people	in kj				calcium	iron
male 15 - 18	11510	55.2	360	109	1000	11.3
female 15 - 18	8830	45.0	276	84	800	14.8
male 19 - 50	10600	55.5	331	100	700	8.7
female 19 - 50	8100	45.0	253	77	700	14.8
pregnant female	8900	81.0	278	84	700	14.8

(a)	(i)	Explain why two 16 year-old males of the same weight might need different amounts of energy.	
			1 mark
	(ii)	Which two types of nutrient provide most of the energy in our diet?	
		1	
		2	2 marks
(b)	(i)	Calculate the difference in the recommended daily intake of calcium for a 15 year-old male and a 30 year-old male.	
		mg	1 mark
	(ii)	Calcium is needed for healthy bones. Explain the difference in the amount of calcium needed each day by a 15 year-old male and a 30 year-old male.	
			1 mark
(c)		k at the table. Explain the difference in the amount of protein needed by a 25 r-old pregnant female and a 25 year-old female who is not pregnant.	

(d)	Iron is needed to make blood.
` ,	Explain why a 15 year-old female might need more iron than a 15 year-old male.
	1 mark
	Maximum 7 marks

Q3.

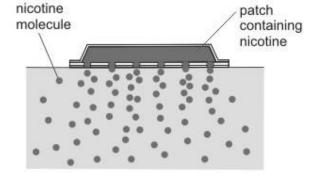
(a) The diagram shows two types of cell in the lining of the airway leading to the lungs.



When a person breathes in cigarette smoke, cilia are damaged and much more mucus is produced. What will be the consequences of this?	
	2 marks

(b) Sharna is addicted to smoking and wants to stop. Instead of smoking she puts nicotine patches on her skin.





nicotine patch on Sharna's skin

a section through a nicotine patch on Sharna's skin

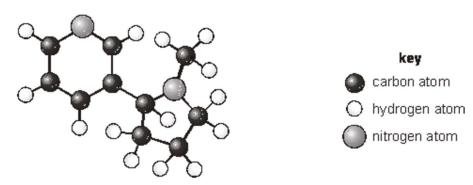
The table shows the mass of nicotine in different patches and the number of weeks each type of patch is used over an eight-week period.

mass of nicotine in each patch (mg)	weeks
21	1-4
14	5-6
7	7-8

Use information in the table to describe how this eight-week course of treatment should help to reduce Sharna's addiction to smoking.	
	2 marks

(c) The diagram below shows a nicotine molecule.

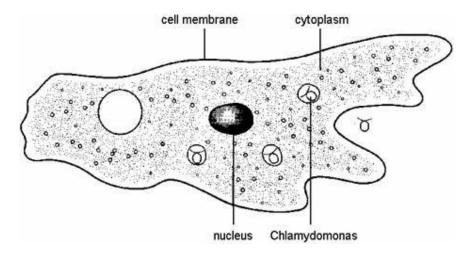
It contains atoms of three elements



i)	Nicotine is a compound. How does the diagram show this?

		(ii)	When nicotine in a cigarette is burned in air, nitric oxide is formed from the nitrogen in the nicotine.	
			Look at the elements in nicotine.	
			Give the names of two other compounds formed when nicotine burns in air.	
			1	
			2	0 1
			maximun	2 marks n 7 marks
0 4	Ī			
Q4	The	diagra I wate	am below shows a single-cell organism called Chlamydomonas. It lives in r.	
			cytoplasm cell wall	
		·	red, light-sensitive spot	
		С	chloroplast nucleus	
		st	carch grain cell membrane	
	Use	the inf	formation in the diagram to help you answer the questions below.	
(a) Give two features of Chlamydomonas which show that it is more like a plant cell than an animal cell.				
		1		
		2		0
	(b)	Chla	amydomonas makes a sugar called glucose.	2 marks
	(5)	(i)	Give the name of the process in which Chlamydomonas makes glucose.	
		(1)	Ove the hame of the process in which offamydomonas makes glacose.	
				1 mark
		(ii)	Chlamydomonas produces starch grains from glucose.	
			Suggest what will happen to the number of starch grains in the cell if Chlamydomonas is kept in the dark.	

(c) The diagram below shows another single-cell organism called Amoeba. It also lives in pond water. Amoeba traps a Chlamydomonas and digests it.



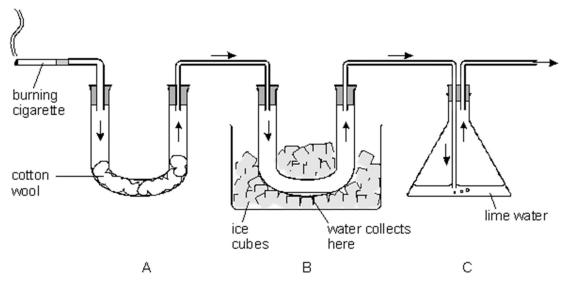
not to scale

Starch is a carbohydrate. Amoeba's digestive enzymes break down the starch in the Chlamydomonas.

Suggest what substance is produced from the starch and what it is used for.	
2 ma	ırks
Maximum 6 ma	rke

Q5.

A teacher set up the following apparatus to separate the chemicals in cigarette smoke. The chemicals pass through the apparatus in the direction of the arrows.



(a)			e of the brown substan	otton wool. This substance caus ce.	,65
					1 mark
(b)	As t	the cigarette burned, v	vater vapour was produ	ced and water collected in B.	
	(i)	Why were ice cubes	needed in B?		
					1 mark
	(ii)	particles of liquid wa		of particles of water vapour and	
		particles of water vapour		particles of liquid water	
					2 marks
(c)	The	lime water in C becar	ne cloudy. What gas tui	rns lime water cloudy?	
					1 mark

Maximum 5 marks

Q6.

This table shows where different pollutants in the air come from.

	source of pollutants			
pollutants	percentage from power stations	percentage from road traffic	percentage from other sources	
sulphur dioxide	72	2	26	
nitrogen oxides	28	51	21	
smoke	6	46	48	
carbon monoxide	1	90	9	

(a)	Which two pollutants in the table above cause acid rain?	
	1	
	2	2 marks
(b)	Lichens are organisms which grow on buildings and trees. They are affected by some pollutants.	
	Use the information in the table to suggest why few lichens survive close to power stations but many survive alongside roads.	
		1 mark
(c)	The two products of complete combustion of petrol are not shown in the table. Give the names of the two products of complete combustion of a fuel such as petrol.	
	1	
	2	2 marks
(d)	Smoke contains particles of carbon. These can burn when there is enough oxygen present. Write a word equation for the reaction of carbon with excess oxygen.	
		1 mark
(e)	Which term best describes the combustion process?	
	Tick one box.	
	decomposition	

oxidation	
pollution	
respiration	

1 mark Maximum 7 marks

Q7.

An alloy is a mixture of elements. The table shows the mass of each element present in 100 g of five different alloys, bronze, solder, steel, stainless steel and brass.

	mass of each element in 100 g of alloy							
alloy	lead (g)	tin (g)	copper (g)	zinc (g)	carbon (g)	iron (g)	chromium (g)	nickel (g)
bronze		4	95	1				
solder	62	38						
steel					1	99		
stainless steel						70	20	10
brass			67	33				

a)	Which alloy in the table above contains an element which is a non-metal?	
		1 mark
b)	Which two alloys in the table contain only two metals ?	
	and	1 mark
c)	Another alloy called nichrome contains only the elements chromium and nickel. 100 g of nichrome contains 20 g of chromium.	
	How much nickel does it contain?	
	g	1 mark

(i)	Why does bronze not rust? Use information in the table above to help you.	
(ii)	Rusting requires water and a gas from the air. Give the name of this gas.	1 marl
		1 mari

maximum 5 marks

Before 1992, two-pence coins were made of bronze.

Steel rusts but bronze does not rust.

Q8.

(d)

In this question you will be assessed on using good English, organising information clearly and using specialist terms where appropriate.

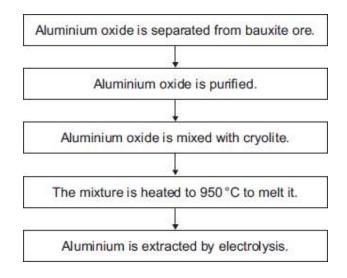
Aluminium is used to make many items.



Car © tridland/iStock

Aluminium is extracted from aluminium ore. Aluminium ore is called bauxite, which is impure aluminium oxide.

The flow chart shows the main steps in the extraction of aluminium from aluminium ore.



Most aluminium is recycled.

•
Aluminium is recycled by melting scrap aluminium at 700 °C.
Use your own knowledge and the information given to answer the question.
Suggest why most aluminium is recycled.

(Total 6 marks)

Q9.

(a) The table gives the names of three different rocks and how they are classified.

name of rock	class of rock	
granite	igneous	
marble	metamorphic	
shale	sedimentary	

In the table below, draw lines to connect the name of each rock to the description of how the rock was formed and then to the correct description of features of the rock.

how the rock was formed	name of rock	features of the rock
layers of mud and tiny dead animals compressed and tumed into rock	granite	large interlocking crystals
magma cooling slowly underground in the Earth's crust	marble	crumbly, layered, containing fossils
limestone changed by heat and pressure	shale	hard, shiny, white with veins of colour, fizzes with acid

3 marks

(b) The diagram shows a section through a limestone cliff.

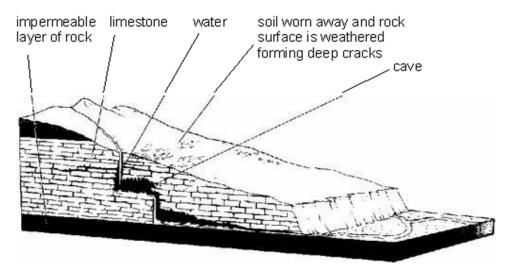


Diagram: Mary Jones, Geoff Jones, David Acaster, Cambridge Coordinated Science: *Chemistry*, 1993

(i) There are caves in the limestone.

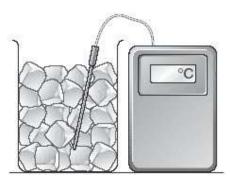
		Explain how chemical weathering causes caves to form in limestone.	
			2 marks
	(ii)	Limestone is a porous rock. Water can enter the spaces in limestone. Describe how this causes physical weathering of the limestone.	
			2 marks
		Мах	kimum 7 marks
Q10.			
(a)		ce probes have shown that there are mountains, dry river-valleys and anoes on Mars.	
		entists believe Mars, like Earth, has all three types of rock: igneous, amorphic and sedimentary. Using the information above, explain why.	
			0 1
4.	_		3 marks
(b)	follov	nperatures on Mars can vary from 10° C in the afternoon to –100° C durin wing night. e how this could affect the rocks on the surface of Mars, and explain why	
			_
		Max	2 marks dimum 5 marks

Q11.

(a) Draw a line from each change of state to the correct name. Draw only **four** lines.

change of state	name	
solid to liquid	evaporating	
liquid to gas	melting	
gas to liquid	condensing	
liquid to solid	freezing	
		3 marks

(b) Kate made some ice cubes from pure water. She used a sensor to measure the temperature of the ice.



What temperature will the sensor show when the ice is melting?
.....°C

1 mark

(c) Kate made some more ice cubes from salt solutions. She used a different amount of salt in each ice cube.

The table shows the temperature at which the ice cubes melted.

mass of salt in each ice cube (g)	temperature ice cube melted (°C)
5	-4
10	-8
15	-11
20	-15

	Look at the table above. As the mass of salt increacube melted?	ased, what happ	pened to the temperature at which the	ice
(-I)		£ t	d d : d d-	1 mark
(d)	in very cold weather a mi	xture of sait and	d sand is spread on roads.	
	Why are salt and sand us Tick the two correct boxe			
Salt	makes the roads white.		Sand dissolves in water.	
Salt	makes water freeze.		Sand increases friction between car tyres and the road.	
Salt	makes ice melt.		Sand makes water freeze.	
			maxir	2 marks num 7 marks

Mark schemes

Q1.

(a)	any one from	
	to prevent it collapsing accept 'protects against collapse'	
	to keep it open 'for protection' is insufficient	
	for support accept 'for strength' accept 'for flexibility'	1 (L5)
(b)	(i) A: oxygen accept 'O₂'	
	B: carbon dioxide accept 'CO₂' both answers are required for the mark	1 (L5)
	(ii) any one from	
	• it is thin	
	it is one cell thick	
	 it is close to the blood supply accept 'there is a diffusion gradient' accept 'it is moist' 	1 (L6)
(c)	(i) any one from	
	it moves mucus accept 'it moves bacteria'	
	it sweeps dust from lungs 'to clear or clean the airways' is insufficient	1 (L6)
	(ii) any one from	
	it paralyses the cilia	
	it stops the cilia working	
	 it clogs the cilia accept 'it destroys them' do not accept 'it kills cilia' 	1 (L5)

	(iii)	nicotine	1 (L6)	[6]
Q2. (a)	(i)	one is more active accept 'one does sport or plays football' accept 'they have different metabolic rates' accept 'one works harder or does more work'	1 (L5)	
	(ii)	carbohydrates answers may be in either order accept a named carbohydrate, eg 'sugar' or 'glucose' or 'starch'	1 (L5)	
		fats	1 (L5)	
(b)	(i)	300	1 (L5)	
	(ii)	any one from		
		 a boy's bones or teeth are still growing accept '15 year-old male is growing or still developing' by 30 the bones have already developed accept '30 year-old male has stopped growing' 	1 (L5)	
(c)	anv	one from	1 (L5)	
(5)	-	a pregnant female supplies the baby with protein accept 'she supplies the baby'		
	• 6	a pregnant female needs protein for herself and the baby accept 'she needs it for herself and the baby'		
	• t	he baby needs protein	1 (L6)	
(d)	any	one from		
	• 6	a 15 year-old female menstruates		
	• 6	a 15 year-old female has periods	1 (L6)	[7]

Q3.

(a) any **two** from

mucus will build up airways will be (partially) blocked accept 'cilia cannot move' accept 'narrower airways' or named part of the airway accept 'breathing will be more difficult' 'cilia do not work as well' is insufficient do not accept 'lungs will be blocked' 2 (L7) any two from reduction in nicotine she gets used to les nicotine accept 'mass of nicotine goes down' accept 'it reduces the craving' or 'it reduces the amount she needs' the patch **or** it provides her with nicotine so she does not need to smoke it provides her with less nicotine than a cigarette 2 (L7) (i) the elements are combined **or** joined **or** bonded accept 'the atoms are joined' accept 'it has a definite composition' 'it has three elements' is insufficient as it is given in the question 1 (L7) (ii) any two from carbon dioxide water carbon monoxide accept 'CO2' accept 'hydrogen oxide' **or** steam **or** H₂O' accept 'CO' accept 'carbon oxide' if neither 'carbon dioxide' nor 'carbon monoxide' is given 2 (L7) [7] answers may be in either order cell wall 1 (L6)

(b)

(c)

Q4.

(a)

chloroplast

	accept 'starch grains'	1 (L6)	
(b)	(i) photosynthesis		
		1 (L6)	
	(ii) it decreases accept 'it gets less' or 'they are used up'		
	do not accept 'they die out'	1 (L6)	
(c)	glucose	1 (L0)	
(0)	accept 'sugar'	1.7.0	
		1 (L6)	
	any one from		
	• for energy		
	for respiration	1 (L6)	
			[6]
Q5.			
(a)	tar	1 (L5)	
(b)	(i) any one from		
	to cool the vapour		
	to condense the vapour		
	accept 'energy is transferred from the water vapour to the ice'		
	100	1 (L5)	
	(ii)		
	0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0		
	<u>66600000</u>		
	a random arrangement a random arrangement of particles most of which of particles most of		
	do not touch which touch each other	2 (L6)	
(c)	carbon dioxide		
(=)	accept 'CO ₂ '	1 (1.5)	
		1 (L5)	[5]
Q6. (a)	sulphur dioxide		
` '	•		

1

1 (L6)

accept 'O2'

1 (L6)

[5]

Q8.

Marks awarded for this answer will be determined by the Quality of Written Communication (QWC) as well as the standard of the scientific response. Examiners should apply a 'best-fit' approach to the marking.

0 marks

No relevant content.

Level 1 (1–2 marks)

A brief reason is given against extraction or for recycling. There is little scientific terminology used.

Level 2 (3-4 marks)

Some reasons are given with clear statements against extraction and or for recycling. Some scientific terminology is used

Level 3 (5-6 marks)

Several reasons are given with a detailed explanation against extraction and for recycling. Scientific terminology is used accurately

examples of chemistry points made in the response

ignore uses and properties of aluminium. Comparative statements count for both methods

extraction:

- limited resources of aluminium oxide
- higher temperatures required
 allow quoted temperatures eg extracted at 950°C
- large amount of energy required
- expensive
- requires mining / quarrying
- process takes longer / has more stages
- produces more carbon dioxide / greenhouse gases

recycling:

- saves resources
- cheaper to recycle
- uses less energy
- only needs to be melted

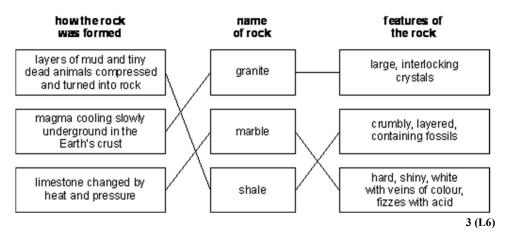
allow quoted temperatures eg melted / recycled at 700°C

- less electricity needs to be used
- less effect on environment
- example of effect on environment eg less destruction of habitats
- · avoids need for disposal / use of landfill
- no need for quarrying
- sustainable

[6]

Q9.

(a) Both the line from the name of the rock to the method of formation and to the features of the rock must be correct for the mark.



(b) (i) acidic water drains through cracks accept 'rain is acidic'

1 (L7)

water reacts with limestone **or** calcium carbonate accept 'limestone dissolves'

1 (L7)

- (ii) any **two** from
 - low temperatures make water freeze in cracks accept 'water freezes in cracks'
 - water expands as it freezes
 - ice expanding breaks up the limestone accept 'ice forces the rock apart'

2 (L7)

[7]

Q10.

(a)	both the type of rock and the evidence for its formation are required
	for each mark

 igneous, because igneous rock comes from volcanoes accept 'igneous, because there are volcanoes'

1 (L7)

any **one** from

- metamorphic, because volcanoes show there were high temperatures accept 'metamorphic, because volcanoes show there had been molten rock'
- metamorphic, because mountains show that the rocks have been under high pressure

accept 'metamorphic, because mountains show there has been land movement'

sedimentary, because there are river valleys and rivers produce sediment

accept 'sedimentary, because there are river valleys' or 'sedimentary, because rivers produce sediment'

2 (L7)

(b) it will break them up **or** split them **or** weather them accept 'it will cause onion skin weathering'

1 (L6)

because they will expand and contract

1 (L6)

[5]

Q11.

solid to liquid evaporating
liquid to gas melting

gas to liquid condensing

liquid to solid freezing

award three marks for **all four** correct lines award two marks for any **three** correct lines award one mark for any **two** correct lines if more than one line is drawn from any change of state, do not credit that change of state

3 (L3)

(b) • 0 °C

accept 'zero' do **not** accept 'nothing'

1 (L4)

(c) • it decreased

accept 'it got colder' 'it dropped to below 0°C' is insufficient any references to time are insufficient

1 (L3)

- (d) Sand increases friction between car tyres and the road. ✓
 - Salt makes ice melt. ✓

if more than two boxes are ticked, deduct a mark for each incorrect box minimum of zero

2 (L4)

[7]